

# Antioxidant Potency of Flavonoids from *Vitex agnus-castus* L. growing in Turkey

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## Summary

Our aim was to investigate the isolation, structure elucidation and determination of radical scavenging properties of flavonoids from *Vitex agnus-castus* L. growing in Turkey. Four flavone glycosides, namely isoorientin (1), 2''-O-trans-caffeoylisoorientin (2), 6''-O-trans-caffeoylisoorientin (3) and luteolin 7-O-glucoside (4) were isolated from the methanol extract of the flowering stems of *V. agnus-castus*. The structure elucidation of the isolated compounds were made by spectroscopic methods (1D and 2D NMR and MS (ESI and DCI)) The radical scavenging activities of compounds 1-4 on DPPH were found to be very high.

**Key Words:** *Vitex agnus-castus*; Verbenaceae; Flavone C and O-glycosides, isoorientin, 2''-O-trans-caffeoylisoorientin, 6''-O-trans-caffeoylisoorientin, Luteolin 7-O-glucoside, Radical Scavenging Activity.

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*Türkiye'de Yetişen Vitex agnus-castus L.'nin Antioksidan Etkili Flavonoitleri*

## Özet

Bu çalışmada Türkiye'de yetişen *Vitex agnus-castus* L.'nin içerdiği flavonoitlerin izolasyonlarının, yapı tayinlerinin yapılması ve serbest radikal süpürücü özelliklerinin saptanması amaçlanmıştır. *Vitex agnus-castus* L.'nin çiçekli dal uçlarının metanol ekstresinden dört flavon glukoziti, izoorientin (1), 2''-O-trans-kafeoilzoorientin (2), 6''-O-trans-kafeoilzoorientin (3) ve luteolin-7-O-glukozit (4) izole edilmiştir. Bileşiklerin yapıları spektroskopik yöntemlerle (1D ve 2D NMR ve MS (ESI ve DCI)) saptanmıştır. 1-4 nolu bileşiklerin DPPH radikal süpürücü aktiviteleri çok yüksek bulunmuştur.

**Anahtar Kelimeler:** *Vitex agnus-castus*; Verbenaceae; Flavon C ve O-glukozitleri, izoorientin, 2''-O-trans-kafeoilzoorientin, 6''-O-trans-kafeoilzoorientin, luteolin 7-O-glukozit, Serbest Radikal Süpürücü Aktivite.

## INTRODUCTION

The genus *Vitex* (Verbenaceae) is represented by two species in the flora of Turkey (1). *Vitex agnus-castus* L. is used not only as a diuretic, digestive, antifungal agent but against anxiety, early birth and stomachache in Anatolia (2,3). *V. agnus-castus* is reported to have medicinal importance in the world. It also has a hormone-like effect. Therefore, it is used for the treatment of premenstrual problems

and hyperprolactinemia (4,5). Investigations on *Vitex agnus-castus* have led to the isolation of iridoids (6-9), diterpenoids (10-12), essential oils (13), ketosteroids (14) and flavonoids (7,15-21).

Previous studies have shown that C-glycosyl flavones are common in genus *Vitex*. Hirobe et al. (15) have isolated four new flavonoids from the root bark of *Vitex*

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*agnus-castus* for the first time in nature, namely luteolin 6-C-(4''-methyl-6''-O-*trans*-caffeoylglucoside), luteolin 6-C-(6''-O-*trans*-caffeoylglucoside), luteolin 6-C-(2''-O-*trans*-caffeoylglucoside) and luteolin 7-O-(6''-*p*-benzoylglucoside) together with four known compounds, namely 4',5-dihydroxy-3,3',6,7-tetramethoxyflavone, luteolin, artemetin and isorhamnetin. In addition, some flavonol and flavone derivatives (apigenin 3, 8-di-C-glycosides, eupatorin, casticin, penduletin, vitexin, isovitexin, isovitexin xyloside, orientin, isoorientin, luteolin 7-O-glucoside) were also isolated from *V. agnus castus* (7,15-17, 20-21).

Previously, we have also reported the isolation and structure elucidation of three new iridoids, namely 6'-O-foliamenthoylmussaenosidic acid (agnucastoside A), 6'-O-(6,7-dihydrofoliamenthoyl) mussaenosidic acid (agnucastoside B) and 7-O-*trans-p*-coumaroyl-6'-O-*trans*-caffeoyl-8-*epi*-loganic acid (agnucastoside C) in addition to four known iridoids (aucubin, agnuside, mussaenosidic acid and 6'-O-*p*-hydroxybenzoylmussaenosidic acid) and one known phenylbutanone glycoside (myzodendrone) from *Vitex agnus-castus* (9). In this study, isolation, structure elucidation and radical scavenging activity of flavonoids isolated from methanol extract of the flowering stems of *V. agnus-castus* were given.

## MATERIALS AND METHODS

### General Experimental Procedures

The UV (MeOH) spectra were recorded on Varian Cary 3E. IR spectra were recorded on a Perkin-Elmer 1600 spectrophotometer. <sup>1</sup>H NMR, <sup>13</sup>C NMR, APT, <sup>1</sup>H-<sup>1</sup>H COSY, HSQC, HMBC and NOESY spectra were determined on Varian Inova 500, Varian U 300, Bruker AMX 300. Chemical shifts were expressed in  $\delta$  values (ppm), relative to TMS. Solvent resonances were used as internal references. ESI/DCI-MS were obtained using a Finnigan LC-Q instrument. Silica gel 60 (0.063-0.200 mm, Merck) and Sephadex LH-20 (Fluka) were used for open column chromatographic separations. TLC analyses were carried out on pre-coated Kieselgel 60 F<sub>254</sub> aluminium sheets (Merck) and compounds were detected by UV fluorescence and spraying 1% vanillin-H<sub>2</sub>SO<sub>4</sub> reagent, followed by heating at 105°C for 1-2 min.

DPPH (2,2-diphenyl-1-picrylhydrazyl) was used for free radical scavenging activity test.

### Plant Material

The flowering stems (flowers, leaves and twigs) of *Vitex agnus-castus* L., collected from Fethiye, Muğla, Turkey in July 2001 were identified by one of the authors, Dr. Ayse Kuruuzum-Uz. A voucher specimen (no 01031) was deposited in Faculty of Pharmacy, Hacettepe University, Ankara, Turkey.

### Extraction and isolation

Dried and powdered flowering stems of *V. agnus-castus* (700 g) were extracted three times with MeOH (3.5 L) at 45 °C and the combined extracts were evaporated under reduced pressure to yield 300 g syrupy residue. The MeOH extract was dissolved in H<sub>2</sub>O and partitioned with CHCl<sub>3</sub> (36 g) followed by *n*-BuOH (74 g). A part of the *n*-BuOH phase (25 g) was fractionated on a silica gel column eluting with a gradient solvent system (CHCl<sub>3</sub>-MeOH) to give nine main fractions (Frs. A-I). Fraction G (3.3 g) was further chromatographed over silica gel column eluting with EtOAc-MeOH-H<sub>2</sub>O (100:5:2 to 100:17:13) to yield eight main fractions (Frs. G<sub>1-8</sub>). Fr G<sub>6</sub> was applied to repeated column chromatographies (CC) over Sephadex LH-20 eluted with MeOH to afford compound 1 (isorientin (=Luteolin 6-C-glucoside), 67 mg) and compound 4 (luteolin 7-O-glucoside, 5 mg). Fr. G<sub>2</sub> was subjected to column chromatography on Sephadex LH-20 using MeOH to give compound 2 (2''-O-*trans*-caffeoylisorientin, 17 mg). Fractionation of Fr. F (10.2 g) by open CC on silica gel using EtOAc-MeOH-H<sub>2</sub>O (100:17:13) yielded subfractions F<sub>1-6</sub>. Fr. F<sub>2</sub> was submitted to Sephadex LH-20 CC (MeOH) to afford pure compound 3 (6''-O-*trans*-caffeoylisorientin, 7 mg).

### DPPH Radical Scavenging Activity Assay

The DPPH assay was carried out using the method described by Cavin et al. (22). The free radical scavenging effect of the isolated compounds 1-4 was assessed by the decoloration of MeOH solution of 2,2-diphenyl-1-picrylhydrazyl (DPPH)

spectroscopically and ascorbic acid were used as standard. Each MeOH solution (230  $\mu$ l) of the tested compounds at various concentrations (100, 50, 25, 10, 5, 1  $\mu$ g/ml) was added to DPPH solution (50  $\mu$ l, 0.022% in MeOH). The mixture was allowed to react for 30 min at room temperature. The absorbance of the solution was read at 517 nm with a spectrophotometer. Radical scavenging activity was determined by comparing the absorbance with that of a blank (100%) containing only DPPH and solvent. The percentage of radical scavenging activity (RSA%) was calculated as follows:  $\text{RSA}\% = ((A_c - A_t) / A_c) \times 100\%$ , where  $A_c$  is the average absorbance of the control and  $A_t$  is the absorbance of the test compounds. All the analyses were made in duplicates.

## RESULTS AND DISCUSSION

In this study, compounds **1-4** were isolated from the methanol extract of the flowering stems of *Vitex agnus-castus* L. by open column chromatographic methods on silica gel and Sephadex LH-20. The structure of the compounds were elucidated as isoorientin (**1**), 2''-O-trans-caffeoylisoorientin (**2**), 6''-O-trans-caffeoylisoorientin (**3**) and luteolin 7-O-glucoside (**4**) by spectroscopic methods (1D and 2D NMR and MS (ESI and DCI)) and by comparison of their physical and spectroscopical data with literature values (18, 23-26).

*Isoorientin* (=Luteolin 6-C-glucoside) (**1**): UV (MeOH)  $\lambda_{\text{max}}$  255, 270, 348. IR (KBr)  $\nu_{\text{max}}$  3390 (OH), 1650 (C=C), 1450 (aromatic ring), 1265, 1083  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz,  $\text{CD}_3\text{OD}$ ):  $\delta_{\text{H}}$  = 7.35 (1H, *dd*,  $J$  = 9.0 / 1.5 Hz, H-6'), 7.32 (1H, *d*,  $J$  = 1.5 Hz, H-2'), 6.92 (1H, *d*,  $J$  = 9.0 Hz, H-5'), 6.50 (1H, *s*, H-3), 6.45 (1H, *s*, H-8), 4.88 (1H, *d*,  $J$  = 9.0 Hz, H-1''), 4.17 (1H, *t*,  $J$  = 9.0 Hz, H-2''), 3.88 (1H, *dd*,  $J$  = 12.0 / 8.0 Hz,  $\text{H}_{\text{B}}$ -6''), 3.74 (1H, *dd*,  $J$  = 12.0 / 5.0 Hz,  $\text{H}_{\text{A}}$ -6''), 3.46 (3H, *m*, H-3'', H-4'', H-5'').  $^{13}\text{C}$  NMR (300 MHz,  $\text{CD}_3\text{OD}$ ):  $\delta_{\text{C}}$  = 184.0 (C-4), 166.2 (C-2), 164.9 (C-7), 162.0 (C-5), 158.7 (C-9), 151.0 (C-4'), 147.0 (C-3'), 123.5 (C-1'), 120.3 (C-6'), 116.8 (C-5'), 114.1 (C-2'), 109.1 (C-6), 105.1 (C-10), 103.9 (C-3), 95.2 (C-8), 82.6 (C-5''), 80.1 (C-3''), 75.3 (C-1''), 72.6 (C-2''), 71.8 (C-4''), 62.9 (C-6''). ESI-MS:  $m/z$  = 471  $[\text{M}+\text{Na}]^+$ ,  $\text{C}_{21}\text{H}_{20}\text{O}_{11}$ .

2''-O-trans-caffeoylisoorientin (**2**): UV (MeOH)  $\lambda_{\text{max}}$  245, 336. IR (KBr)  $\nu_{\text{max}}$  3390(OH), 1650(C=C), 1578, 1450 (aromatic ring), 1267, 1079  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz,  $\text{CD}_3\text{OD}$ ):  $\delta_{\text{H}}$  = 7.36 (1H, *d*,  $J$  = 16.0, H-7'''), 7.29 (1H, *d*,  $J$  = 1.5 Hz, H-2'), 7.28 (1H, *dd*,  $J$  = 9.0 / 1.5 Hz, H-6'), 6.90 (1H, *d*,  $J$  = 2.0 Hz, H-2'''), 6.84 (1H, *d*,  $J$  = 9.0 Hz, H-5'), 6.80 (1H, *dd*,  $J$  = 8.0 \ 2.0 Hz, H-6'''), 6.69 (1H, *d*,  $J$  = 8.0 Hz, H-5'''), 6.44 (1H, *s*, H-3), 6.39 (1H, *br. s*, H-8), 6.05 (1H, *d*,  $J$  = 16.0 Hz, H-8'''), 5.66 (1H, *br. s*, H-2''), 5.10 (1H, *d*,  $J$  = 10.0 Hz, H-1''), 3.92 (1H, *dd*,  $J$  = 12.5 / 2.0 Hz,  $\text{H}_{\text{B}}$ -6''), 3.79 (1H, *dd*,  $J$  = 12.5 / 5.5 Hz,  $\text{H}_{\text{A}}$ -6''), 3.72 (1H, *t*,  $J$  = 9.0 Hz, H-3''), 3.60 (1H, *t*,  $J$  = 9.0 Hz, H-4''), 3.50 (1H, *m*, H-5'').  $^{13}\text{C}$  NMR (300 MHz,  $\text{CD}_3\text{OD}$ ):  $\delta_{\text{C}}$  = 183.8 (C-4), 168.3 (C-9'''), 166.1 (C-2), 164.8 (C-7), 161.7 (C-5), 158.7 (C-9), 150.9 (C-4'), 149.4 (C-4'''), 146.9 (C-7''', C-3'\*), 146.6 (C-3''''\*), 127.6 (C-1'''), 123.4 (C-1'), 122.9 (C-6'''), 120.3 (C-6'), 116.7 (C-5'), 116.4 (C-5'''), 115.0 (C-2'''), 114.8 (C-8'''), 114.1 (C-2'), 107.9 (C-6), 104.9 (C-10), 103.8 (C-3), 95.2 (C-8), 82.8 (C-5''), 77.9 (C-3''), 73.8 (C-2''), 73.2 (C-1''), 71.7 (C-4''), 62.8 (C-6''); \* signals are interchangeable. ESI-MS:  $m/z$  = 633  $[\text{M}+\text{Na}]^+$ ,  $\text{C}_{30}\text{H}_{26}\text{O}_{14}$ .

6''-O-trans-caffeoylisoorientin (**3**): UV (MeOH)  $\lambda_{\text{max}}$  245, 336. IR (KBr)  $\nu_{\text{max}}$  3390(OH), 1650(C=C), 1578, 1450 (aromatic ring), 1267, 1079  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz,  $\text{CD}_3\text{OD}$ ):  $\delta_{\text{H}}$  = 7.55 (1H, *d*,  $J$  = 16.0, H-7'''), 7.36 (1H, *dd*,  $J$  = 8.5 / 1.5 Hz, H-6'), 7.34 (1H, *d*,  $J$  = 1.5 Hz, H-2'), 7.02 (1H, *d*,  $J$  = 2.0 Hz, H-2'''), 6.92 (1H, *dd*,  $J$  = 8.0 \ 2.0 Hz, H-6'''), 6.88 (1H, *d*,  $J$  = 8.5 Hz, H-5'), 6.75 (1H, *d*,  $J$  = 8.0 Hz, H-5'''), 6.53 (1H, *s*, H-3), 6.48 (1H, *s*, H-8), 6.28 (1H, *d*,  $J$  = 16.0 Hz, H-8'''), 4.90 (1H, *d*, H-1''), 4.52 (1H, *dd*,  $J$  = 12.0 / 2.0 Hz,  $\text{H}_{\text{B}}$ -6''), 4.34 (1H, *dd*,  $J$  = 12.0 / 5.5 Hz,  $\text{H}_{\text{A}}$ -6''), 4.23 (1H, *t*,  $J$  = 9.0 Hz, H-2''), 3.65 (1H, *m*, H-5''), 3.51 (2H, *m*, H-3'', H-4'').  $^{13}\text{C}$  NMR (300 MHz,  $\text{CD}_3\text{OD}$ ):  $\delta_{\text{C}}$  = 183.9 (C-4), 169.2 (C-9'''), 166.1 (C-2), 165.2 (C-7), 162.1 (C-5), 158.7 (C-9), 151.1 (C-4'), 149.6 (C-4'''), 147.1 (C-7'''), 147.0 (C-3'), 146.7 (C-3'''), 127.7 (C-1'''), 123.4 (C-1'), 123.1 (C-6'''), 120.3 (C-6'), 116.8 (C-5'), 116.4 (C-5'''), 115.1 (C-2'''), 114.9 (C-8'''), 114.1 (C-2'), 108.8 (C-6), 105.0 (C-10), 103.8 (C-3), 95.2 (C-8), 79.9 (C-3'', C-5''), 75.4 (C-1''), 72.4 (C-2''), 71.8 (C-4''), 65.0 (C-6''). ESI-MS:  $m/z$  = 633  $[\text{M}+\text{Na}]^+$ ,  $\text{C}_{30}\text{H}_{26}\text{O}_{14}$ .

Luteolin 7-O-glucoside (**4**): UV (MeOH)  $\lambda_{\text{max}}$  255, 336. IR (KBr)  $\nu_{\text{max}}$  3389 (OH), 1654 (C=C), 1445 (aromatic ring), 1070  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (300 MHz,  $\text{CD}_3\text{OD}$ ):  $\delta_{\text{H}}$  = 7.43

(1H, dd,  $J = 8.0 / 2.0$  Hz, H-6'), 7.40 (1H, br. s, H-2'), 6.88 (1H, d,  $J = 8.0$  Hz, H-5'), 6.77 (1H, d,  $J = 2.0$  Hz, H-8), 6.71 (1H, s, H-3), 6.43 (1H, d,  $J = 2.0$  Hz, H-6), 5.06 (1H, d,  $J = 7.0$  Hz, H-1''), 3.71 (1H, br. d,  $J = 10$  Hz, H<sub>B</sub>-6''), 3.48 (1H, br. dd,  $J = 14.5 / 5.5$  Hz, H<sub>A</sub>-6''), 3.46 (1H, m, H-5''), 3.30 (1H, m, H-3''), 3.28 (1H, t,  $J = 7.5$  Hz, H-2''). <sup>13</sup>C NMR (300 MHz, CD<sub>3</sub>OD):  $\delta_c = 181.8$  (C-4), 164.5 (C-2), 162.9 (C-7), 161.1 (C-5), 156.9 (C-9), 150.2 (C-4'), 145.8 (C-3'), 121.1 (C-1'), 119.1 (C-6''), 115.9 (C-5'), 113.4 (C-2'), 105.3 (C-10), 103.0 (C-3), 99.9 (C-1''), 99.5 (C-6), 94.7 (C-8), 77.1 (C-5''), 76.4 (C-3''), 73.1 (C-2''), 69.6 (C-4''), 60.6 (C-6''). DCI-MS:  $m/z = 449$  [M+H]<sup>+</sup>, C<sub>21</sub>H<sub>20</sub>O<sub>11</sub>.

Compounds **1-4** were isolated as yellow amorphous powders. Their UV and IR spectra were characteristic of a flavone system.

Molecular formula of compound **1** was determined as C<sub>21</sub>H<sub>20</sub>O<sub>11</sub> by the NMR spectra and ESI-MS. In the ESI mass spectrum of the compound **1**, peak [M+Na]<sup>+</sup> was observed at  $m/z$  471. In the <sup>1</sup>H NMR spectrum of the compound **1**, H-6' and H-2' gave doublet ( $J=9$  Hz) and singlet at  $\delta$  7.35 and 7.32, resp. H-5' proton signal was observed at 6.92 ppm as a doublet ( $J=9$  Hz). H-3 and H-8 protons were at  $\delta$  6.50 (1H, s) and  $\delta$  6.45 (1H, s) respectively. Anomeric proton (H-1'') appeared at  $\delta$  4.88 as doublet with  $J=7$  Hz to indicate  $\beta$ -glucosyl unit. Comparing with the reported data, the <sup>1</sup>H NMR and <sup>13</sup>C NMR data are in agreement with those of isoorientin in articles (23-26). Therefore, the structure of the compound **1** was determined as isoorientin.

The ESI-MS of compound **2** and **3** which exhibited [M+Na]<sup>+</sup> at  $m/z$  633 showed that they had the same molecular formula (C<sub>30</sub>H<sub>26</sub>O<sub>14</sub>). The NMR spectra of **2** and **3** showed the presence of a flavone skeleton with one caffeoyl and one C-glucosyl group. The C-glucose was assigned to C-6 from the cross peak between C6/H-1'', C7-H-1'' and C5-H-1'' in the HMBC spectrum. The caffeoyl groups were assigned to C-2'' for compound **2** and C-6'' for compound **3** from the cross peak between C-9'''- H-2'' and H-6'', resp. in the HMBC spectrum and the *trans*-geometry was elucidated from the coupling constants between H-7''' and H-8''' ( $J = 16.0$  Hz). The NMR spectra of **2**

were similar to those of **3**, but the chemical shifts of C-2'' and C-5'' were shifted downfield in comparison with those of **3**, and those of C-1'', C-3'', and C-6'' were shifted upfield. Also, the chemical shifts of H-1'', H-2'' and H-3'' in **2** were shifted downfield in comparison with those of **3**, but those of H-5'' and H-6'' were shifted upfield. These results demonstrated that there were caffeoyl groups at C-2'' in **2** and C-6'' in **3**. Consequently, the structures of compounds **2** and **3** were determined as 2''-*O*-*trans*-caffeoylisorientin and 6''-*O*-*trans*-caffeoylisorientin, respectively (15, 23-26).

The molecular formula of compound **4** was determined as C<sub>21</sub>H<sub>20</sub>O<sub>11</sub> by DCI-MS, which exhibited a molecular ion peak at  $m/z$  449 [M+H]<sup>+</sup>. The <sup>1</sup>H- and <sup>13</sup>C-NMR spectral data of compound **4** indicated a luteolin bearing one glucose moiety. The site of glycosylation in **4** was determined to be the C-7 position based on the chemical shifts of H-6 and H-8 in comparison to that reported for luteolin (**26**) and from the cross peak between C-7 and H-1'' ( $\delta_c$  162.9 and  $\delta_H$  5.06) in the HMBC spectrum for compound **4**. The chemical shift value of the anomeric proton and carbon atom for compounds **4** as well as the C resonances of the hydroxyl group indicated that the  $\beta$ -glucose unit were connected at C-7. In comparison with the literature, data of luteolin were found to be in good agreement with those recorded for luteolin-7-*O*-glucoside in our study. Therefore, we conclude that compound **4** is luteolin 7-*O*-glucoside (23-26).

In addition, we tested the isolated flavonoids for their free radical scavenging activities by comparison with ascorbic acid as reference. DPPH radical scavenging activities of the compounds **1-4**, isolated from methanol extract are shown in Table 1. Compounds **1-4** were found to have significant free radical scavenging activity.

## CONCLUSION

*V. agnus-castus* is reported to have medicinal importance. Previous studies of the secondary metabolites from *V. agnus-castus* have shown that C-glycosyl flavones are common in genus *Vitex* (27).

**Table 1.** Scavenging activities of the compounds 1-4 on DPPH radical.

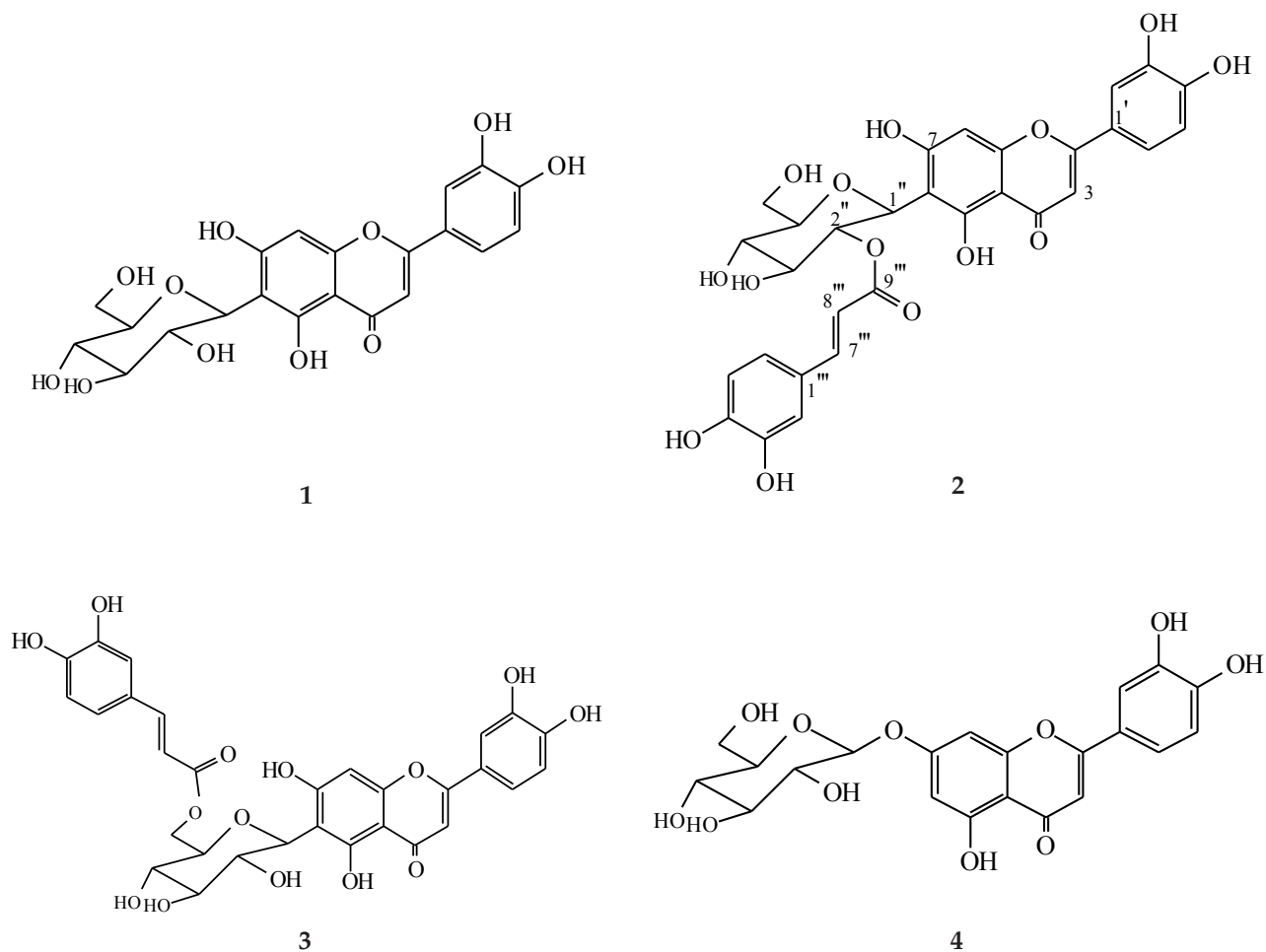
Compounds	IC <sub>50</sub> <sup>*</sup>
1	16.9
2	23.6
3	23.4
4	21.8
Ascorbic acid <sup>a</sup>	12.0

<sup>\*</sup>IC<sub>50</sub> values were calculated from regression lines using six different concentrations in duplicate.

<sup>a</sup> Positive control

We now report the results of chemical and biological examination of methanol extract of the flowering stems of *V. agnus-castus*. And this is the first report of the isolation of flavonoids from the flowering stems of *V. agnus-castus* growing in Turkey.

Oxidative stress has been linked to cancer, aging, atherosclerosis, ischemic injury, inflammation and neurodegenerative diseases (Parkinson's and Alzheimer's). Flavonoids may provide a protection against these diseases. *V. agnus-castus* is rich in flavonoid derivatives, which have high antioxidant properties.



**Figure 1.** Flavonoid glycosides of *Vitex agnus-castus*.

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